# Chemistry 342-Organic Chemistry II Summer 2015

Instructor: Dr. Rui ZhangClassroom: Snell 3107Office: TCCW 443Meeting Time: MTWF 9:00 am -12:00 amOffice Tel: 745-3803Office Hour: Walk-in or by appointmentEmail: rui.zhang@wku.eduText: "Organic Chemistry, 11th edition" by SolomonsSupplemental Text: "ACS Organic Chemistry Exam (Study Guide)" and "Organic Chemistry (II) asa Second Language" by David R. KleinPrerequisites: CHEM 340 with a grade of "C" or better. Corequisites: CHEM 343

#### **Important Dates**

Monday, July 6: Classes begin.

Friday, July 17, Last day to drop with a W. Last day to change a course from credit to audit. Friday, July 31, 9 am-11:00 am: Final ACS Exam in Snell 3107.

**Course Description:** CHEM 342 is the second semester of a two-semester sequence, which will focus on providing broader insights into the reactions and mechanisms of organic compounds and their chemical reactivity. Fundamentals of general chemistry and organic chemistry, nomenclature, as well as spectroscopy methods will be reinforced.

### **Course Outline and Sequence**

- Chapter 10: Radical Reactions
- **Chapter 11**: Alcohols and Ethers
- Chapter 12: Alcohols from Carbonyl Compounds
- Chapter 13: Conjugated Unsaturated System:
- Chapter 14: Aromatic Compounds
- Chapter 15: Reactions of Aromatic Compounds
- Chapter 16: Aldehydes and Ketones: The Carbonyl Group
- Chapter 17: Carboxylic Acid and Their Derivatives
- **Chapter 18:** Reactions at the *alpha* Carbon of Carbonyl Compounds
- Chapter 19: Condensation and Conjugate Additions of Carbonyl Compounds

#### Grading

This course will be graded on a *curve*. Approximate grade cut-offs will be announced after each exam. You will be graded on a relative, subjective scale in this course, and the points will be distributed as follows (subject to minor modification):

Homework ( $10 \times 10$ )	100
Quiz Section ( $10 \times 10$ )	100
Midterm Exams $(4 \times 100)$	400
ACS Exam (mandatory)	100
Total	<b>700</b> points

**Exams:** Midterm Exams 1 and 3 (1 hour test time) will be <u>given at regular class meeting time</u> in Snell 3107 on the following specified dates listed below. Exam 2 and 4 will be <u>take-home</u>. *The exam dates may be subject to change based on actual teaching pace.* 

Examination Dates:	Monday, July 13, Exam 1 ( <b>Ch 10-12</b> ); In class
	Friday, July 17, Exam 2 ( <b>Ch 13-15</b> ); Take Home
	Friday, July 24, Exam 3 ( <b>Ch 16, 17</b> ); In class
	Thursday, July 30, Exam 4 ( <b>Ch 18, 19</b> ); Take Home
	Friday, July 31, ACS Exam (cumulative)

*No exams will be regraded during this semester and no partial credit will be given for questions of nomenclature and reaction.* I will make every effort to ensure that your work is graded carefully and fairly, and that your scores are entered accurately in my records

**Quizzes**-10 quizzes will be given throughout the course. They will last only ca. 10 minutes. Quizzes will be administered either at the start of lecture or take-home format. There are no make up's for quizzes. **10**-Points may be earned for each quiz.

**Extra Credit:** Significant improvement (> 20 points) in the course of study, reflecting from your latest Test's score (10 points). Bonus questions (10 points) from each midterm exam.

#### Blackboard

We will use the Blackboard web-based system in the course. Teaching notes, tutorial materials, assignments, practice problems, answer keys, and grade distributions will be posted at that site in PDF files.

## **Attendance Policy**

Poor attendance reflects a poor scholastic attitude. Excessive and non-excused absences (4 or more) will result in a **failing** grade. Three absences will drop your grade by one letter. A sign-in sheet will be passed around for your signature in each class. No excuse is accepted for more than four absences.

## **Study Tips**

Because over 10 million organic compounds exist, memorizing the structure, properties, and reactivity of all of them would be almost impossible. Luckily, **a few** fundamental ideas underlie all organic reactions. By **understanding** these themes and trends (*not by memorizing them!*), you should be able to rationalize unfamiliar reactions and mechanisms through analogy.

Prepare for class! Read the chapter scheduled to be covered before coming to lecture. Understanding will not occur on the first exposure to new material, therefore if the lecture presentation is your second sighting of the concepts, your understanding will be much better. You will also be prepared to ask questions about complex ideas.

- Lectures are important! Skipping lecture will guarantee failure. The lecture and the textbook serve complementary roles. The lectures are meant to highlight concepts, simplify difficulties and expand on implications. In lecture you should be attentive. A mature student will transcribe not only what the instructor writes on the board, but also what the instructor says.
- Apply your knowledge! Doing problems is the quickest way to discover whether your understanding is incomplete or incorrect.
- Study regularly and actively! Consistent study of chemistry is essential for success. Simply studying two days before the exam will make failure likely.
- Ask questions! There is no such thing as a bad question. You pay good money to attend this school, so take advantage of the instructor's understanding and availability.

#### **Students with Disabilities:**

In compliance with university policy, students with disabilities who require accommodations (academic adjustments and/or auxiliary aids or services) for this course must contact the Office for Student Disability Services in DUC A-200 of the Student Success Center in Downing University Center. Please DO NOT request accommodations directly from the professor or instructor without a letter of accommodation from the Office for Student Disability Services.

Although I will make every effort to follow this syllabus exactly, I reserve the right to make changes as I deem necessary throughout the course.